## Spring 2009 CH302 Worksheet 10 – Organic Chemistry

- 1. How many structural isomers does  $C_4H_{10}$  have? Draw them.
- 2. Name them.
- 3. How many structural isomers does  $C_7H_{16}$  have? Draw them.
- 4. Name them.
- 5. Circle and name the functional groups in these famous compounds.

## Penicillin G

Norethindrone - active ingredient in Enovid (the first "pill")

Capsaicin - "heat" causing molecule in chili peppers

6. Provide the IUPAC name.

- 7. Draw the following:
  - a. 3,3,5-Triethylheptane
  - b. 2-Methylcyclohexanone
- 8. Draw the following:
  - a. 5,5-Dimethyl-1-hexene
  - b. 2,3,4-trimethyl-4,7-dipropylnonan-1-al
  - c. 5-ethyl-1,3,6-heptatriene.
- 9. Name the following.
  - a. CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>
  - b.  $CH_2(CH_2CH_3)_2$
  - c. CH<sub>2</sub>(OH)CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH(CH<sub>2</sub>CH<sub>3</sub>)CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>
- 10. Proteins are biological polymers made of amino acids. What type of linkage forms these polymers? Draw the product that would result from these two amino acids reacting together to form a peptide. What type of reaction is this?

$$H_3$$
CO
 $CH_3$ 
 $H_3$ 
 $CH_3$ 

Alanine methyl ester

*N*,*N*-dimethylglycine

11. What levels of structure organization are possible for proteins and what do these each describe?

- 12. Draw and name the four DNA bases.
- 13. Which are the guanines and which are the pyrimidines?
- 14. What is responsible for the helical structure of DNA? Draw a circle around the H-bonds between base pairs.
- 15. What is the monomer and formula in the famous polymer that comprises irrigation pipes (PVC, polyvinyl chloride)?
- 16. What types of polymers are there (4 general classes)? Draw examples with A,B notations.
- 17. What is a nucleoside? What is a nucleotide?
- 18. What is a fatty acid? What is the difference between saturated and unsaturated?
- 19. What type of reaction is shown below?

20. What type of reaction generates this ether?